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⑮ 発明の名称 半導体基板の洗浄方法

⑯ 特 願 昭59-125760

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明 細 書

1. 発明の名称

半導体基板の洗浄方法

2. 特許請求の範囲

(1) 半導体基板を有機酸にオゾン/酸素をバブルした溶液中に浸漬し、処理することを特徴とする半導体基板の洗浄方法。

3. 発明の詳細な説明

〔技術分野〕

本発明は半導体基板の洗浄方法に関するものである。

〔従来技術とその問題点〕

LSI 製造において、ウェハの洗浄工程は各プロセス間にゴミ、汚れを除去する数多くの処理回数が必要とし、製品の歩留り向上に重要な技術となっている。この洗浄工程で使われている主な液としてはアルカリ系と酸系に大別される。公知の洗浄液としてはアンモニア水-過酸化水素系、又は塩酸-過酸化水素系(通称 RCA 洗浄ともいう)。さらに、硫酸-過酸化水素系、硫酸-硝酸系等の

混酸溶液が用いられている。溶液は通常60～150℃に加熱され、時には超音波を併用することもある。

しかしながら、アンモニア-過酸化水素系や塩酸-過酸化水素系では、過酸化水素がすぐに H_2O と発生の O に分解してしまうので、酸化力が劣化し、洗浄力が持続せず間欠的に過酸化水素を補充する必要がある、工程としては複雑になる。硫酸-過酸化水素系も同じ理由で不利である。この硫酸-過酸化水素系や硫酸-硝酸系等は廃液として処理する場合の問題がある。それは硫酸を廃棄する場合、水と反応して100℃以上になり安全上危険があり、このため大量の水を必要とし、又硝酸を使う場合は、窒素酸化物等が発生し公害上規制される等、大量に使うことはできないという不利な点がある。すなわち、処理費用が高くなる。

又上記の酸-過酸化水素系に代る処理方法としての公知例として特公昭52-012063号の半導体基板のレジスト膜除去方法がある。この方法は硫酸を用いて過酸化水素の代りにオゾンを用いる方法

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(54) Name of Invention: Cleaning Method for Semiconductor Substrate
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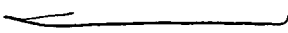
Specifications

1. Name of Invention

Cleaning Method for Semiconductor Substrate

2. Claims

(1) A cleaning method for a semiconductor substrate is characterized by a process in which a semiconductor substrate is dipped into an organic acid that is fed with bubbles containing ozone/oxygen.



3. Detailed Description of the Invention

<Technical Field>

This invention pertains to a cleaning method for a semiconductor substrate.

<Prior Art & Related Problems>

During LSI production, wafers must be cleaned several times between processes in order to remove dust and dirt, and a technique for doing so has become important from the standpoint of improving production yields. The primary solutions used in such a cleaning process are generally classified into alkali and acid type solutions. Examples of well-known cleaning solutions include aqueous ammonia – hydrogen peroxide and hydrochloric acid – hydrogen peroxide (also commonly referred to as RCA cleaning) types. In addition, mixed acid solutions are used such as sulfuric acid – hydrogen peroxide and sulfuric acid – nitric acid types. Solutions are normally heated to a temperature of 60 – 150°C and sometimes undergo ultrasonic wave treatment during heating.

However, in the cases of ammonia – hydrogen peroxide and hydrochloric acid – hydrogen peroxide types, since the hydrogen peroxide quickly breaks down into H_2O and nascent-state O, the deterioration in oxide strength causes inconsistent cleaning strength. Therefore, it becomes necessary to periodically replenish the solution with hydrogen peroxide, resulting in a complicated process. Sulfuric acid – hydrogen peroxide type solutions are also unfavorable for the same reason. Problems exist in cases where these sulfuric acid – hydrogen peroxide and sulfuric acid – nitric acid types are used as waste solutions during treatment. In cases where sulfuric acid is discarded, there is a safety concern when it reacts with water and reaches a temperature of 100°C or more. For this reason, a large volume of water is needed. Furthermore, when sulfuric acid is used, nitrogen oxides are generated which need to be regulated due to pollution standards, and for this reason, sulfuric acid cannot be used in large amounts. In short, the problems noted above result in high treatment costs.

In addition, Patent No. S52-012063 describes a method for removing resist films from semiconductor substrates in a well-known treatment method that takes the place of the aforementioned acid – hydrogen peroxide type solutions. According to this method, sulfuric acid is used along with ozone, which takes the place

of hydrogen peroxide. However, according to this method, a large volume of water (up to a tenfold dilution) is required in order to discard the sulfuric acid, which presents a problem from the standpoint of safety as well as cost.

<Purpose of the Invention>

The purpose of this invention is to present a treatment method in which it is possible to maintain a high level of safety as well as a low cost.

<Constitution of the Invention>

The method for cleaning semiconductor substrates as described in this invention is characterized by a process in which an organic acid is used as a cleaning solution and bubbles containing ozone/oxygen are fed into the solution tank in order to remove dirt from the surface of silicon wafers.

<Principle/Operation of the Invention>

Among the various types of organic acids, formic acid is a strong acid that contains reduction properties. On the other hand, acetic acid is a weak acid in which the hydrogen within the -COOH group becomes displaced once the acetic acid reacts with the various heavy metals (M), resulting in the formation of a -COOH metallic salt.

Furthermore, by feeding bubbles containing ozone into the solution tank, organic contaminants become oxidized by the ozone, making it possible to remove these contaminants through decomposition.

In other words, this invention seeks to conduct a cleaning operation in which heavy metals located on top of semiconductor substrates (wafers) create formic acid salt or acetic acid salt, and organic contaminants are decomposed through the use of ozone.

<Embodiments>

The following is an explanation of this invention in which a prior example is used for comparison.

First, an embodiment of this invention will be explained.

According to this invention, the cleaning operation is conducted as follows. A cleaning tank is filled with an organic acid (such as formic acid, acetic acid, etc.), and the solution is heated (to a temperature of $100 - 150^{\circ}\text{C}$), after which a semiconductor substrate is dipped into the solution. Ozone/oxygen bubbles are fed from the bottom of the tank and clean the substrate as they hit up against the surface.

The tank used in this invention is made of Teflon, and the solution temperature is controlled through the use of an immersion type quartz heater. The ozone/oxygen bubbles are fed into the tank from the bottom using a gas distributor made of Pyrex. The same type of tank is used to conduct cleaning with a prior form of RCA cleaning solution comprised of $\text{NH}_4\text{OH} - \text{H}_2\text{O}_2 - \text{H}_2\text{O}$ (1:1:5) at a solution temperature of 80°C for 10 minutes.

In addition to using the same tank described above, the method of this invention calls for the use of a wafer from the same lot, and said wafer is to be dipped into a solution of formic acid or acetic acid. While the wafer is within the tank, the solution is heated to a temperature of $100 - 150^\circ\text{C}$. In addition, ozone/oxygen bubbles are fed through the bottom of the tank so that cleaning can be conducted for a period of 10 minutes.

The wafer that is cleaned using the $\text{NH}_4\text{OH} - \text{H}_2\text{O}_2 - \text{H}_2\text{O}$ solution according to the prior method, and the wafer that is cleaned using the formic acid – ozone – O_2 solution according to this invention both undergo a steam treatment (in an environment of $\text{H}_2 - \text{O}_2$ at 950°C for 10 minutes), after which lifetime measurements are conducted using a non-contact method. These results are shown in Table 1 along with the V_{FB} (Flat Band Voltage) figures, which are evaluated using the MOS C–V method.

Table 1

Cleaning Method	Lifetime (μ sec)	V_{FB} (V)
Prior Method (NH_4OH)	7	-0.9
Method of this Invention (Ozone/Formic Acid)	20	-1.0

<Effect of the Invention>

As shown in Table 1, the cleaning method described in this invention has a recombination lifetime value that is approximately three times that of the RCA cleaning solution from the prior method, which indicates a superior cleaning effect. In addition, the V_{FB} (Flat Band Voltage) figure obtained through the MOS C–V method shows a level of -1.0V , which is not significantly different from the result shown for the prior method.

Based on the above, the cleaning method described in this invention provides a cleaning effect that is equal to or greater than that of the prior method, thus sufficiently achieving the purpose stated above.

In addition, the problem noted above with respect to the generation of heat due to the reaction between water and sulfuric acid or nitric acid is prevented, thus this invention creates very few problems from the standpoint of pollution. This is due to the fact that formic acid and acetic acid can be easily decomposed into CO, CO₂, H₂O, etc., at high temperatures (up to 150°C), and this decomposition is further advanced through the use of ozone in the treatment process. As a result, this invention poses no problem with regard to discarding the solution since it can be easily dissolved in water. This not only makes it possible to provide industrial safety as well as a rational manufacturing operation, it also makes it possible to reduce the level of cost required to conduct the treatment process.

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